

Glutathione Reductase: A lucky turn to remediate plant's oxidative stress

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Novelty statement:

glutathione reductase is an essential enzyme in plants that contributes to cellular redox homeostasis. Its activity is vital for maintaining the balance between oxidized and reduced glutathione and is also responsible for the regeneration of reduced glutathione (GSH) from its oxidized form, glutathione disulfide (GSSG). The primary function of glutathione reductase is to maintain the cellular pool of reduced glutathione, which is an essential antioxidant molecule. The enzyme is found in different subcellular compartments, including the cytosol, chloroplasts, mitochondria, and peroxisomes, reflecting its diverse functions within the plant cell. The activity of glutathione reductase in plants can vary depending on various factors such as developmental stage, tissue type, and environmental conditions. The present review gives a comprehensive overview of the importance of this enzyme.

Abstract

Abiotic conditions have a significant negative impact on the growth, maintenance, and ultimately yield of the plant, which causes significant losses in terms of food crisis. The excessive production of hazardous reactive oxygen species (ROS), such as superoxide ion, hydrogen peroxide, and hydroxyl radicals, is the cause of oxidative stress, which is connected to basically all abiotic stressful situations. As a result of these reactions, the plants suffer cellular harm and molecular deterioration, which ultimately results in plant death. In plants under stressful conditions, a series of reactions (enzymatic and non-enzymatic) has started that enhance plant resistance to such conditions. A major enzyme of the antioxidant system is glutathione reductase (GR) because it is essential for maintaining the sulfhydryl (-SH) group. It maintains the GSH depletion through the glutathione-ascorbate pathway. Reduced glutathione also involves the removal of reactive oxidants which are necessary byproducts of metabolic activities so it is recognized as one of the most critical cellular antioxidants. On the other hand, glutathione reductase primarily keeps the GSH pool in its reduced form. Changes in glutathione redox status can be caused by a variety of biotic or abiotic stressors that impact the rate of ROS formation and detoxification.

Keywords: Glutathione reductase, Glutathione, ROS, abiotic stress

Due to photosynthesis, plants serve as a main producer in the food chain and as a result, consumers benefit. These beneficial but static creature is facing environmental stresses. Plants have developed a variety of defence mechanisms against it. Two types of factors biotic such as diseases, pathogens and abiotic such as intense light, high temperature, cold, water stress, and salt stress factors affect plant growth and metabolism (Maksymiec, 2007).

Salinity, heavy metals, water stress, high temperature, cold, and UV-B radiation are examples of abiotic stress factors that can similarly alter the structure of plants. Positive (such as tolerance) or negative (such as decreased photosynthetic activity, growth inhibition, accelerated senescence, or damage to the plant parts) impacts can be seen depending on the severity of the stress (Maksymiec, 2007; Mittler, 2006). These effects are typically caused by redox imbalances and excessive ROS production in the cells. ROS ($O_2^{\cdot-}$, OH^{\cdot} , HO_2^{\cdot} , H_2O_2) are normally present in various cell compartments (Karuppanapandian et al., 2008; Mafakheri et al., 2010; Mittler, 2002; Mittler et al., 2004; Torres et al., 2002; Vellosillo et al., 2010) but under stress and as a result of the loss of redox homeostasis, they may build up and cause damage to a variety of molecules,

including DNA, proteins, fats, photosynthetic pigments, and others (Arora et al., 2002; Dat et al., 2000; Jaleel et al., 2009).

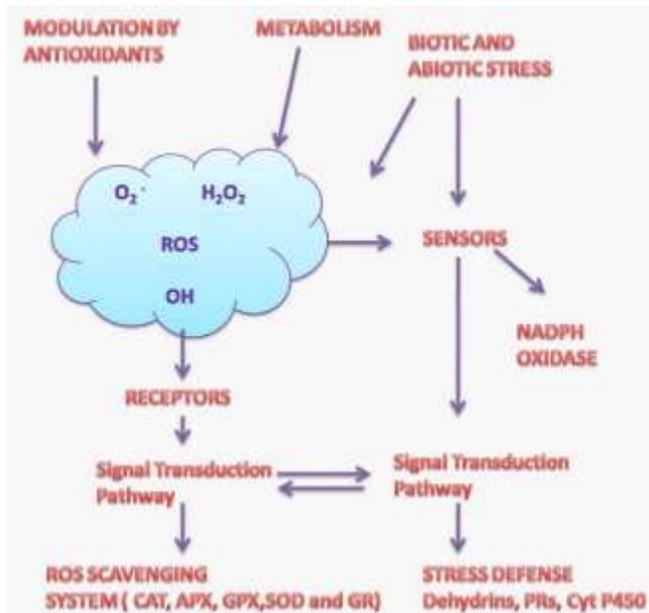


Figure 1. ROS scavenging by antioxidants in plants under biotic and abiotic stress.

It is believed that a vast gene network known as the "ROS gene network" controls ROS generation and scavenging, enabling this duality in function to exist in plants (Mittler et al., 2004). To generate the energy required for their developmental activities, plants need oxygen (O₂). Ground state oxygen is converted during typical cellular metabolism to water (H₂O) and reactive oxygen species, which include O₂⁻, H₂O₂, HO²·, OH· and ¹O₂ (Halliwell, 2006; Mittler, 2002; Scandalios, 2005). It is assumed that 1- 2% of the O₂ absorbed by plants is diverted to the ROS formation in distinct intercellular sites (Blokhina et al., 2003)). ROS are produced from O₂ either through energy transfer or electron transfer processes. The reaction chain of ROS production initially requires an energy input, but the following phases are exothermic and spontaneous. Absorption of surplus energy by O₂ can also result in the production of ¹O₂, which is a more reactive molecule than O₂ (Halliwell, 2006; Mittler, 2002). ¹O₂ is formed in chloroplasts as a result of the photosensitization of chlorophyll (Chl) molecules under UV stress (Rao & Reddy, 2008).

$O_2^{\cdot-}$ is produced by a single electron reduction of O_2 which is moderately reactive and transient ROS. As a result, $O_2^{\cdot-}$ is unable to pass through biomembranes and easily dismutates into H_2O_2 . Protonation in aqueous solutions converts $O_2^{\cdot-}$ into HO_2^{\cdot} . By removing hydrogen atoms from lipid hydroperoxides, HO_2^{\cdot} can pass across bio-membranes and start auto-oxidation of lipids (Halliwell & Gutteridge, 2015). H_2O_2 is a moderately reactive, long-lasting molecule that can permeate short distances away from its source. Enzymes can be inactivated by H_2O_2 by oxidizing their thiol groups. By oxidizing the thiol groups on enzymes, H_2O_2 can render them inactive. H_2O_2 might freely pass through membranes, allowing it to disperse harm and perhaps work as a secondary messenger in the plant's hormonal response to stress (Halliwell, 2006; Møller et al., 2007). This shows that H_2O_2 (ROS) is dual in its nature. H_2O_2 is one of the radicals that start lipid peroxidation(LP) since it can also produce OH^{\cdot} , the ROS family member that is highly reactive (Halliwell & Gutteridge, 2015; Lee et al., 2007).

Stressful situations have the potential to change the equilibrium between ROS generation and removal in plant cell organelles, even though reactive oxygen species are a byproduct of regular cellular metabolism (Apel & Hirt, 2004; Vellosillo et al., 2010). Superoxide radical ($O_2^{\cdot-}$), hydroxyl radical (OH^{\cdot}), hydroperoxyl radical (HO_2^{\cdot}), hydrogen peroxide (H_2O_2), alkoxy radical (RO^{\cdot}) proxy radical (ROO^{\cdot}), singlet oxygen (1O_2) and excited carbonyl (RO^*) are among the reactive oxygen species that are detrimental to plants (Dismukes et al., 2001; Karuppanapandian et al., 2008; Manoharan et al., 2005; Vellosillo et al., 2010).

All biomolecules, severely harming biological components, Genome instability and abnormalities are targeted by ROS and usually lead to cell death and permanent metabolic losses (Karuppanapandian et al., 2011). As in (figure.2) severe oxidative stress can cause cells to go through fatal response pathways such as apoptosis, necrosis, and possibly other types of cell death pathways that can eventually result in apoptosis (Awasthi et al., 2015)

Additionally, ROS can serve as a supplementary messenger for controlling the number of biological and developmental processes as well as in pathogenic resistance i.e., the HR: hypersensitivity response in plants (Foyer & Noctor, 2005; Guan & Scandalios, 2000; Mittler et al., 2004; Pei et al., 2000).

It is not expected that OH^{\cdot} radical itself act as a signaling molecule. However, the results of OH^{\cdot} reactions can trigger signaling procedures. Cells effectively avoid OH^{\cdot} by sequestering

the catalytic metals into metallochaperones (small proteins responsible for transporting metals) (Halliwell, 2006; Møller et al., 2007). Hydroxide (OH^\cdot) can react with all biological compounds, including pigments, proteins, fats, and DNA, as well as nearly every cell element. Because these extremely reactive ROS cannot be scavenged, so their excess synthesis leads to programmed cell death (PCD) (Karuppanapandian et al., 2011; Manoharan et al., 2005; Vranová et al., 2002).

The reactive oxygen species are removed by numerous antioxidant defence mechanisms in stable circumstances (Foyer & Noctor, 2005; Navrot et al., 2007). In plant cells, both the production of reactive oxygen species and their scavenging are controlled processes, and the plant's reaction is determined by the balance of oxidative and antioxidative capacity. Under a stress-free environment, the antioxidant defence system provides enough protection against ROS; but, under stress conditions, ROS generation overcomes plant scavenging capacity, which causes environmental stress (Apel & Hirt, 2004).

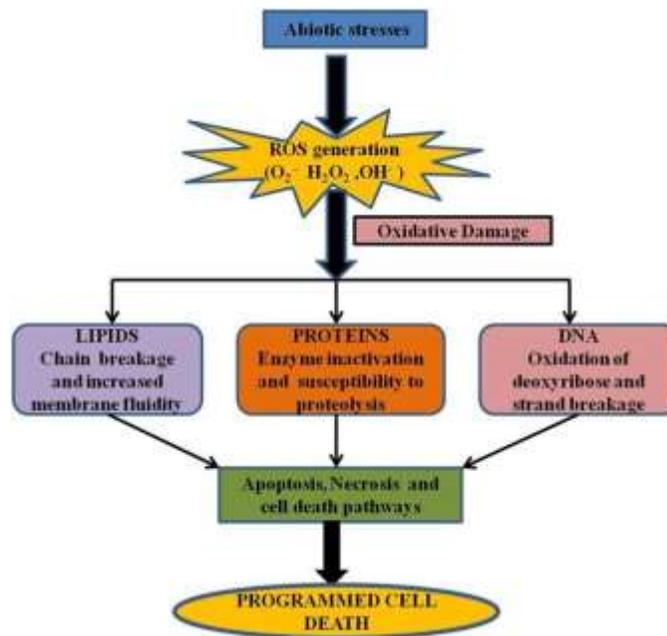


Figure 2. Reactive oxygen species (ROS) generation under abiotic stress (Awasthi et al., 2015)

Like all living organisms, Plants produce a variety of organic substances and follow strategies to regulate biological ROS production or the increased ROS levels under stressful conditions (Yousuf et al., 2012). These molecules, which are termed antioxidants, are present in almost all cellular organelles and function in a systematic manner (Gill & Tuteja, 2010). Antioxidants are crucial for the best optimal functioning of plant cells as they serve as the first protective barrier against damage from oxidants (Bartels & Sunkar, 2005; Gill & Tuteja, 2010; Miller et al., 2010; Rajput et al., 2016). Antioxidants have a role in the scavenging pathways of ROS as like the Halliwell-Asada (or ascorbate-glutathione) cycle in the chloroplast, the water-water cycle in the mitochondria, the peroxisomes, the apoplast, the cytosol and the cycle of peroxisomal glutathione peroxidase. Because of their inert nature, plants have developed a complex network of cellular antioxidants made up of both enzymatic and non-enzymatic components that are essential for defending against a range of stressors (Rajput et al., 2021).

The antioxidant defence mechanism prevents oxidants from harming other biological elements such as enzymes, nucleic acids, and unsaturated proteins and lipids. As a result, the scientific community has been showing a lot of interest in the defence mechanism of plants (Alscher et al., 1997; Dumont & Rivoal, 2019). Non-enzymatic components like free amino acids, α -tocopherols, alkaloids, flavonoids, carotenoids, phenolic compounds, ascorbic acid (AA), and glutathione (GSH). Enzymatic substances are ascorbate peroxidase (APX), catalase (CAT), superoxide dismutase (SOD), monodehydroascorbate reductase (MDHAR), peroxidases (POX), glutathione peroxidase (GPX), glutathione reductase (GR), glutathione S-transferases (GST), and dehydroascorbate reductase (DHAR) (Figure.4) (Bhardwaj et al., 2021; Maximiano & Franco, 2022; Rajput et al., 2015). Glutathione reductase includes in the second line of defence mechanism. The term "scavenging antioxidants" is frequently used to describe this class of antioxidants. They disrupt chain propagation reactions and scavenge active radicals to prevent chain start. By giving free radicals an electron, they can neutralize or scavenge them in the process to transform into new but less harmful free radicals. By using other antioxidants of this class, these "new radicals" are quickly neutralized and rendered completely harmless (Ighodaro & Akinloye, 2018)

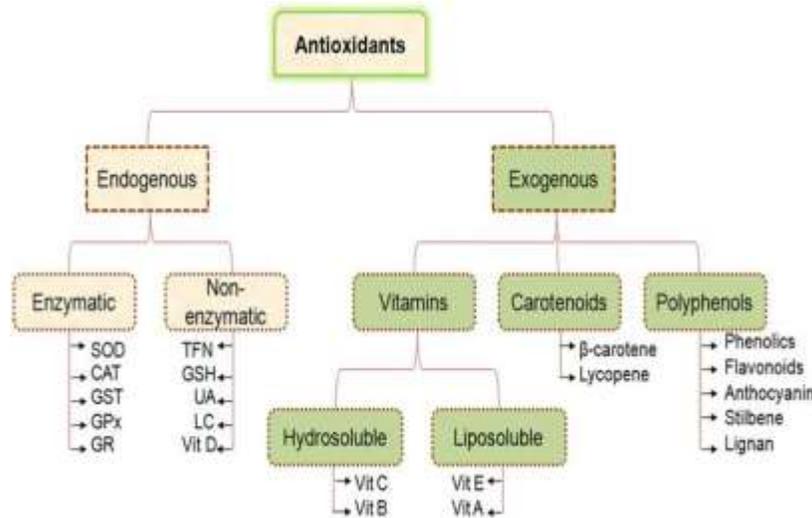


Figure 3. Schematic representation of general antioxidants (Bhardwaj et al., 2021)

Antioxidant enzymes perform their functions in plants under stress conditions through a cascade manner. SOD dismutates oxygen (O^{-1}) during stressful settings into O^2 and H_2O_2 , CAT converts the H_2O_2 into water and molecular oxygen (O_2), and POX works in the extracellular space to scavenge H_2O_2 . Plant GPX catalyses the conversion of H_2O_2 and HO^{-2} to water and lipid alcohols, respectively, by using thioredoxin as an electron donor. The conversion of oxidised glutathione (GSSG; dimeric) to reduced glutathione (monomeric form) is catalyzed by glutathione reductase (GR) (Rajput et al., 2021). At this stage, glutathione reductase plays an important role by maintaining the GSH/GSSG ratio and provides GSH to glutathione peroxidase (GPX). The GSH pool (GSH/GSSG ratio) and reducing environment in the cell are maintained by GR, which is important for the active operation of proteins (Creissen et al., 1994; Edwards et al., 1990).

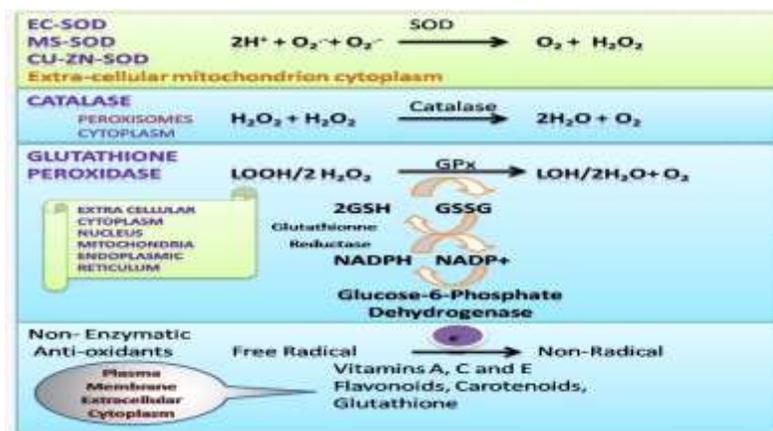


Figure 4. Removal of ROS species by the activation of the antioxidant cascade.

These enzymes also play a crucial role in the development and germination of plants by regulating cellular and subcellular processes like mitosis, morphogenesis, ageing, and apoptosis. They also serve to protect various cell constituents from harm. Additionally, they play a role in various mechanisms, including cell differentiation, growth and division, regulation of senescence and sulphate transport, detoxification of xenobiotic, metabolites complexation, control of enzymes metabolic activity, synthesis of proteins and nucleotides, phytochelatins, and genes functioning that are responsive to stress (Abdel Latif, 2011; Liu et al., 2014; Maximiano & Franco, 2022; Mullineaux & Rausch, 2005; Sairam et al., 2011).

Glutathione is an oxidoreductase where NADPH is used as a cofactor that conducts both the oxidation and reduction processes. Because glutathione reductase is found in chloroplasts, cytosol, and mitochondria, it has been demonstrated that in the photosynthetic tissue of plants, the isoform of chloroplast is responsible for more than 80% of its activity (Ashraf, 2009; Edwards et al., 1990; Romero-Puertas et al., 2006; Stevens et al., 2000). Glutathione reductase generates a homodimer that is linked to Flavin adenine dinucleotide (FAD). The majority of glutathione reductases in plants are homodimers with one FAD per monomer and molecular weights between 100 and 150 kDa. When thiols are not present, GR generally forms tetramers and larger forms. Generally, glutathione reductase forms tetramers and bigger forms in the absence of thiols. Under cellular circumstances, the enzyme's product GSH keeps the enzyme in its dimeric state despite the catalytic activity of these larger forms (Yousuf et al., 2012). Instead of reagents or products, both pH and temperature determine the composition of the higher form, thus GR dimers can be converted into tetramers or either greater constituting states. GR catalytic activity is regulated by this mechanism (Rao & Reddy, 2008). Glutathione reductase (GR) is a part of the antioxidant defence mechanisms of plants because it participates in both enzymatic and non-enzymatic oxidation reduction activities within the cell.

Utilizing NADPH as a cofactor, glutathione reductase changes oxidised glutathione (GSSG) into reduced glutathione (GSH) (Edwards et al., 1990; Romero-Puertas et al., 2006; Stevens et al., 2000). One mole of NADPH is needed to turn one mole of GSSG into one mole of GSH. The catalytic process consists of two steps: first, NADPH is used to reduce the flavin group, and an oxidative disulfide bond is reduced to produce a cysteine and a thiolate anion after the

flavin is first oxidised. Thiol disulfide interchange reactions are used to minimize GSSG in the second stage (Ghisla & Massey, 1989). If GSSG does not reoxidize the reduced enzyme, there can be a reversible inactivation. GR functions in a ping-pong manner during the reduction of GSSG to GSH, NADPH binds to FAD, transfers a hydride to it, and then dissociates before glutathione can bind (Rao & Reddy, 2008). Furthermore, GR controls the proportion of reduced to oxidized glutathione and provides GSH with glutathione peroxidase and dehydroascorbate reductase. NADPH, H^+ provides the reduction power for GR, but GR dissipates this power, which raises the $NADP^+/NADPH, H^+$ ratio (Hasanuzzaman et al., 2012). In cells where GSH builds up, plants become more tolerant to stress and their glutathione reductase activity rises. In terms of GR catalytic processes, it is important to note that the amount of available substrate heavily influences the GR redox interconversion, in contrast to the reduced GR form, the oxidised GR form exhibits better stability because it can withstand divalent metal ions (Rao & Reddy, 2008). GR, which also significantly contributes to reactive oxygen species detoxification and GSH regeneration, helps in the tolerance to environmental stress conditions in plants (Hasanuzzaman et al., 2012; Mirza et al., 2010). In addition to providing a resistance to stress, enhanced GR activity has the capability of changing the redox condition of the essential transport system of electron constituents. The maintenance of oxidized to reduced glutathione ratio in the cells of the plants and GSH recycling play a significant role in GR's ability to resist stress (Pang & Wang, 2010; Rao & Reddy, 2008).

Many physiological processes depend on the regulation of a large GSH/GSSG ratio by glutathione reductase (GR), and a decline in this ratio could be used as an indication of osmotic stress. Additionally, the reduced to-oxidized glutathione ratio is essential for controlling several procedures related to plant growth and signalling pathways. GSH content and subcellular distribution, in addition to redox status, are important elements in the regulation of redox signalling and homeostasis (Sabetta et al., 2017). Plant homeostasis is disturbed due to ROS production in excess under stressful climate change situations.

Glutathione (GSH) is a common thiol tripeptide with a low molecular weight and antioxidant molecule that contains Sulphur which is essential for managing plant growth, productivity and tolerance to stress both biotically and abiotically. Under stress conditions, glutathione readily develops in plant cells. Oxidized (GSSG) and reduced (GSH) are the two types of glutathione. under ideal circumstances, glutathione is mostly found in its reduced form (GSH), which has a

free thiol group. Disulfide glutathione(GSSG) is created when two molecules of reduced glutathione (GSH) form a disulfide bond(Sabetta et al., 2017). As reduced glutathione can remove reactive oxidants which are necessary byproducts of metabolic activities so it is recognized as one of the most critical cellular antioxidants. As a result, in both environmental and biological stresses in plants, GSH play an important role, where its function is to eliminate ROS and hence reduce the amount of oxidative damage (Foyer & Noctor, 2005). Moreover, GSH is involved in the amplification of ROS signals in plants via interactions with stress hormones (Han et al., 2013).

The abundant metabolite GSH in plants is known to have a role in signal transduction and directly scavenges OH^{\bullet} and IO^2 . It may also protect enzyme thiol groups (Foyer & Noctor, 2005). GSSG accumulation in plants frequently occurs under stressful conditions and is associated with a rise in the total glutathione pool, which appears to be mostly brought on by GSSG accumulation (Mhamdi et al., 2010; Smith et al., 1984; Willekens et al., 1997). under stress conditions, oxidative activities overcome glutathione reductases (GR) ability to reduce glutathione. Increased glutathione disulfide (GSSG) concentration activates -glutamylcysteine synthetase (-ECS), resulting in a rise in the total glutathione pool (Figure.1). Moreover, the cytoplasm and nucleus, which are delicate subcellular regions, are protected from excessive GSSG buildup by the compartmentalization of GSSG in vacuoles. In response to oxidative stress, GSH biosynthesis can rise because thiol and GSH formation are triggered at the translational and post-synthetic levels in response to oxidative stress, GSH biosynthesis may increase (Gromes et al., 2008; Hicks et al., 2007; Queval et al., 2009). Many GSH functions are controlled by these reversible redox processes. GSH synthesis depends on glutamylcysteine synthetase and glutathione synthetase both of these are dependent on ATP, which are respectively encoded by the nucleus genes having GSH1 and GSH2. Plastids are the first stage in the synthesis of reduced glutathione. The second step, on the other hand, can occur in either the plastids or the cytosol (Sabetta et al., 2017).

The reduced to oxidized glutathione ratio in different cell constituents can be used as a reliable indicator of oxidative stress and this is the reason for increase in oxidized glutathione in some portions, such as the vacuole (Noctor et al., 2013). In plants under normal conditions, GSH and GR play the most important function in the H_2O_2 scavenging pathway in chloroplasts (Halliwell & Foyer, 1978). The high reduction status of the cellular pool of GSH is an essential

factor. Glutathione reductase primarily keeps the GSH pool in its reduced form, the activities of which are dependent on the primary electron carrier and NADPH (Edwards et al., 1990; Halliwell & Foyer, 1978). On the other hand, tolerance level to any kind of stress is increased due to the presence of glutathione reductase and reduced glutathione. The reduced to oxidized glutathione proportion as well as the overall GSH level have an impact on the redox capacity of glutathione and on balancing of interaction between targets for sensitive proteins and oxidative signals (Meyer et al., 2007). Due to the high sulfhydryl (SH) concentration of GSH, it guards against cellular component damage by scavenging free radicals. As a result, it acts as a sensitive indicator of cell growth and function (Tanwir et al., 2021). Different circumstances where oxidant generation takes place can cause glutathione to deviate from its highly reduced state that impacts the rate of ROS formation and detoxification (Gómez et al., 2004; Gupta et al., 1991; Vanacker et al., 2000).

GSH functions as a scavenger, limiting severe cellular oxidation. GSH can also generate mixed disulfides when it reacts with various thiols. Many GSH functions are controlled by these reversible redox processes. Glutathione is involved in cell cycle regulation, redox signalling, enzymatic activity and also sense changes in cellular reduced and oxidized glutathione levels. As GSH's antioxidant and signaling roles are linked so it needs an enzyme like glutathione reductase (GR) (Sabetta et al., 2017). Glutathione reductase not only regulates the subcellular redox environment but is also involve in plant reproduction which is crucial for cells to survive (Trivedi et al., 2013).

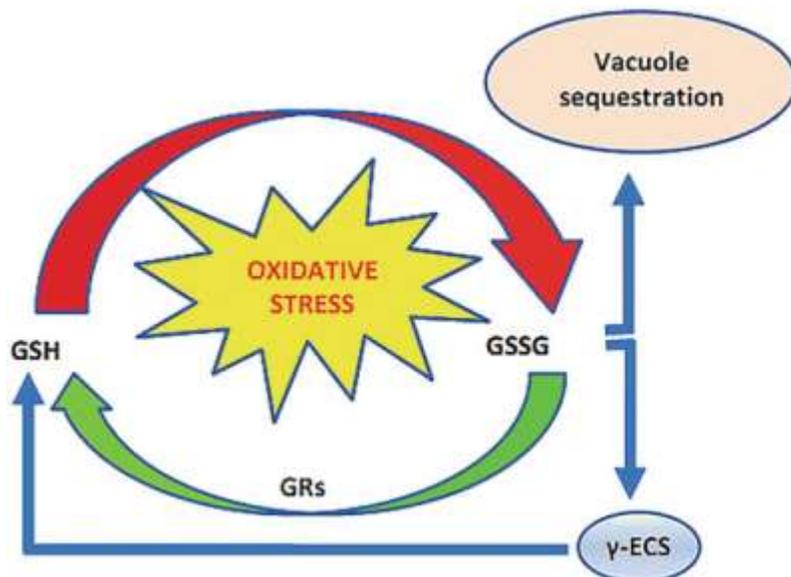


Figure 5. Glutathione content and redox state changes as a result of oxidative stress (Sabetta et al., 2017).

The glutathione redox potential, which regulates the interactions between oxidant indicators and sensitive protein receptors, can be influenced not only by variations in the reduced-to-oxidized glutathione ratio but also by variations in the overall reduced glutathione content (Meyer et al., 2007). Under oxidative stress condition in plants if the glutathione redox pool is not maintained properly then plants undergoes oxidative damage which leads to the death of the plant. Therefore, regulation of the GSH pool and tolerance to oxidative stress for plants is impossible without glutathione reductase (GR).

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